

## A Simple and Efficient Synthesis of Bufalin<sup>1</sup>

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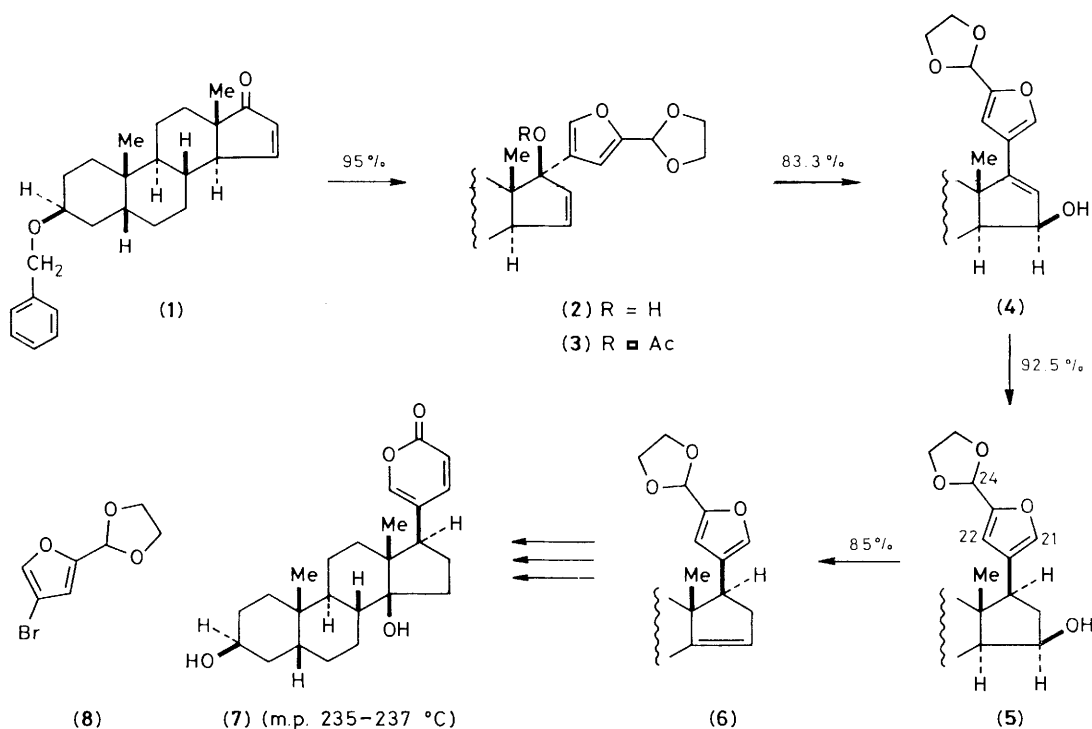
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A simple high-yield preparation of bufalin (**7**) from the steroid (**1**) (derived from testosterone) by an adaptation of our 'furan methodology' for cardenolides is reported.

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Some time ago we reported an efficient synthesis of digitoxigenin *via* furyl-containing intermediates.<sup>2</sup> At an appropriate stage of the synthesis the furyl group was converted into the

unsaturated lactone by a high-yield oxidation reaction. We now disclose a direct preparation of bufalin by an adaptation of this strategy. Lengthy, many-stage syntheses of this natural



product have been reported previously,<sup>3</sup> but a truly simple and efficient preparative method does not seem to have been found yet.

Treatment of the bromoacetal (8)<sup>4</sup> with *n*-butyl-lithium in diethyl ether followed by addition of the protected ketone (1)<sup>2</sup> at  $-70^{\circ}\text{C}$  gave the foamy allylic alcohol (2) in a yield of 95%. Acetylation of (2) with acetic anhydride-pyridine and reflux of the crude acetate (3) in aqueous acetone in the presence of  $\text{CaCO}_3$  yielded 83.3% of the allylic rearrangement product (4) (m.p.  $140\text{--}141^{\circ}\text{C}$ , recrystallised from ether).

Hydrogenation of (4) (ethanol-Pd/ $\text{CaCO}_3$ ) gave the pure amorphous dihydro-derivative (5) in a yield of 92.5%. Elimination of the  $15\text{-}\beta$  hydroxy-group in (5) with mesyl chloride in pyridine yielded finally 85% of the unsaturated compound (6) (m.p.  $141\text{--}142^{\circ}\text{C}$  from ether-hexane);  $^1\text{H}$  n.m.r. ( $\text{CDCl}_3$ )  $\delta$  0.68 [s, 3H, 18-H], 0.97 [s, 3H, 19-H], 3.70 [m,  $W_{\frac{1}{2}}$  ca. 7 Hz, 3-H], 3.88–4.20 [m, O- $\text{CH}_2\text{-CH}_2\text{-O}$ ], 4.47 [s,  $-\text{CH}_2\text{-Ph}$ ], 5.23 [m,  $W_{\frac{1}{2}}$  ca. 6 Hz, 1H, 15-H], 5.83 [s, 1H, 24-H], 6.33 [br. s, 1H, 22-H], 7.18 [br. s, 1H, 21-H], and 7.28 [br. s, 5 arom. H].

We prepared compound (6) some time ago from digitoxigenin and converted it into bufalin by a photo-oxidative furan ring opening followed by a modification of the functional

group system.<sup>5</sup> This simple high-yield process was now repeated and the product was fully verified with the totally synthetic material. The synthetic bufalin (7) was identical in all respects with the natural compound.

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